Report of...

### **Compliance Emission Testing**

performed for...

## Plastic Plate, LLC. Kraft Avenue Plant

Kentwood, Michigan

#### RECEIVED

JUN 1 6 2017 AIR QUALITY DIV.

Multiple Sources

on

April 18 thru 21, 2017

021.30

Network Environmental, Inc. Grand Rapids, MI

# JUN 1 6 2017

#### I. INTRODUCTION

Network Environmental, Inc. was retained by Plastic Plate, LLC to perform compliance emission sampling on multiple sources located at their Kraft Avenue facility in Kentwood, Michigan. The purpose of the study was to document compliance with Michigan Department of Environmental Quality, Air Quality Division, Renewable Operating Permit MI-ROP-N7374-2015 and Source-Wide Permit to Install MI-PTI-N7374-2015.

The following is a list of the sources, applicable emission limits and the compounds tested:

Stack ID	Emission Limits	Compound Sampled
SVK1	DCP: 1.5 Lbs/Hr	1,3 Dichloro-2-propanol
SVK4	Methanol: 9.0 Lbs/Hr Formaldehyde: 1.1 Lbs/Hr NaOH: 0.22 Lbs/Hr	Methanol, Formaldehyde & Sodium Hydroxide
SVK6	Nickel: 0.19 Lbs/Hr Formaldehyde: 0.04 Lbs/Hr	Nickel & Formaldehyde
SVK7	Nickel: 0.04 Lbs/Hr Formaldehyde: 0.04 Lbs/Hr NaOH: 0.33 Lbs/Hr	Nickel, Formaldehyde & Sodium Hydroxide
SVK8	Total Cr: 0.003 Lbs/Hr and 0.006 Mg/M <sup>3</sup>	Total Chromium

The sampling was performed by Stephan K. Byrd, R. Scott Cargill, Richard D. Eerdmans and David D. Engelhardt of Network Environmental, Inc. over the period of April 17-21, 2017. Assisting in the study was Ms. Karen Baweja of Lacks Industries and the operating staff of the facility. Mr. Tom Gasloli and Ms. April Lazzaro of the Michigan Department of Environmental Quality, Air Quality Division, were present to observe the testing and source operation.

#### **II. PRESENTATION OF RESULTS**

II.1 TABLE 1 NICKEL EMISSION RESULTS SEMI BRIGHT (SVK6) & BRIGHT (SVK7) EXHAUSTS PLASTIC PLATE, LLC KENTWOOD, MICHIGAN APRIL 19 and 20, 2017						
Semi Bright (SVK6) Sample #	Time	Air Flow Rate DSCFM	Concentration Mg/M <sup>3</sup>	Mass Emission Rate Lbs/Hr		
1 (4/20/17)	8:24-9:28	32,743	0.0137	0.00168		
2 (4/20/17)	10:06-11:11	32,355	0.0098	0.00119		
3 (4/20/17)	11:38-13:03	32,315	0.0143	0.00173		
Avera	ge	32,471	0.0126	0.00153		
Bright (SVK7) Sample #						
1 (4/19/17)	8:10-9:16	30,738	0.0070	0.00081		
2 (4/19/17)	9:57-11:04	32,222	0.0063	0.00076		
3 (4/19/17)	11:37-12:43	31,351	0.0064	0.00075		
Avera	Average 31,437 0.0066 0.00077					

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#### II.2 TABLE 2 FORMALDEHYDE EMISSION RESULTS PLASTIC PLATE, LLC KENTWOOD, MICHIGAN APRIL 19,20 and 21,2017

Source	Date	Sample∖#	Time	Air Flow Rate DSCFM	Mass Emission Rate Lbs/Hr
Electroless Copper	4/19/17	1	8:10-9:10	26,587	0.1654
(SVK4)		2	9:24-10:24		0.1634
		3	10:40-11:40		0.2050
		Average			0.1779
		1	8:15-9:15	32,058	0.0155
Semi-Brite Nickel (SVK6)	4/21/17	2	9:23-10:23		0.0129
		3	10:129-11:29		0.0126
		Average			0.0137
		1	8:00-9:00		0.0141
Brite Nickel (SVK7)	4/20/17	2	9:15-10:15	30,639	0.0161
		3	10:24-11:24		0.0131
		Average			0.0144

#### II.3 TABLE 3 METHANOL EMISSION RESULTS ELECTROLESS COPPER (SVK4) EXHAUST PLASTIC PLATE, LLC KENTWOOD, MICHIGAN APRIL 19, 2017

Sample Time	Air Flow Rate DSCFM	Concentration Mg/M <sup>3</sup>	Mass Emission Rate Lbs/Hr
1 8:10-9:10		6.527	0.6498
2 9:24-10:24	26,587	2.296	0.2286
3 10:40-11:40		4.476	0,4455
Average		4.433	0.4413

#### II.4 TABLE 4 DCP EMISSION RESULTS CONDITIONER (SVK1) EXHAUST LACKS ENTERPRISES KENTWOOD, MICHIGAN APRIL 21, 2017

IN A SAMINA SA PARA A SA PARA A PA	ow Rate Concentration Mass Emission CFM Mg/M <sup>3</sup> Lbs/Hr	マイマ しきものい とうせい
1 7:44-8:44	0.454 0,0090	
2 8:52-9:52 5	313 2.978 0.0592	
3 9:56-10:56	3.292 0.0655	
Average	2.241 0.0446	5

#### II.5 TABLE 5 SODIUM HYDROXIDE EMISSION RESULTS PLASTIC PLATE, LLC KENTWOOD, MICHIGAN APRIL 19 and 20, 2017

Source	Date	Sample #	Time	Air Flow Rate DSCFM	Mass Emission Rate Lbs/Hr
Electroless Copper		1	8:10-9:10		0.0265
(SVK4)	4/19/17	2	9:24-10:24	26,587	0.0272
		3	10:40-11:40		0.0110
		Average			0.0216
		<b>1</b>	8:00-9:00		0.0444
Brite Nickel (SVK7)	4/20/17	2	9:15-10:15	30,639	0.0249
		3	10:24-11:24		0.0243
		Average		1 <u> </u>	0.0312

#### II,6 TABLE 6 TOTAL CHROME EMISSION RESULTS PLASTIC PLATE, LLC KENTWOOD, MICHIGAN APRIL 18, 2017

Source	Sample	Time	Air Flow Rate DSCFM	Concentration Mg/M <sup>3</sup>	Mass Emission Rate Lbs/Hr
Chrome	1	7:26-9:30	36,430	0.00164	0.000223
Plate	2	10:06-12:09	36,186	0.00223	0.000303
	3	12:38-14:41	36,384	0.00161	0.000219
	Average	•	36,333	0.00183	0.000248
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#### **III. DISCUSSION OF RESULTS**

The emission results are presented in Tables 1 through 6 (Section II.1 through II.6).

#### **IV. SAMPLING AND ANALYTICAL PROTOCOL**

All of the sampling locations met the optimum requirements of U.S. EPA Reference 1. All exhaust stack dimensions and all of the point locations can be seen in Appendix F. Twelve points (six per port) were used for all of the air flows and isokinetic sampling.

**IV.1 Nickel** - The nickel emission sampling was conducted in accordance with U.S. EPA Method 29 (multiple metals train), Figure 1 is a schematic diagram of the Method 29 sampling train. Each sample was sixty (60) minutes in duration and had a minimum sample volume of thirty (30) dry standard cubic feet. The samples were collected isokinetically on quartz filters, and in a nitric acid/hydrogen peroxide solution.

The samples were recovered and refrigerated until they were analyzed. The filters and nozzle/probe rinses (front half) were combined with the impinger catch of nitric acid/hydrogen peroxide solution and were analyzed for nickel by Inductively Coupled Argon Plasma (ICAP)/Mass Spectrometer (MS). All the quality assurance and quality control procedures listed in the methods were incorporated in the sampling and analysis.

**IV.2 DCP and Methanol** - The methanol and DCP determinations were performed in accordance with EPA Method 308. Teflon probes were used to extract the exhaust gas from the exhausts. Silica Gel sorbent tubes were used to collect the methanol and DCP samples. The sampling trains were operated with vacuum pumps with calibrated critical orifices. Two midget impingers were used ahead of the tubes. Each impinger containined approximately 15mls of DI water. One sample spike was run for each compound. The spikes were liquid and were added to the DI water impinger for the spike trains. The orifices were calibrated at approximately 1000 cc/min. Three (3), sixty (60), minute samples were collected from the exhausts for each compound. Figure 2 is a schematic diagram of the DCP and Methanol sampling train.

The silica gel tubes and impinger contents were recovered and refrigerated until analyzed. The tubes were desorbed and the impinger contents and tubes were analyzed by GC/FID in accordance with the method for methanol or DCP. All quality assurance and quality control requirements specified in the method were incorporated in the sampling and analysis. In addition, a spiked duplicate train was run during one of the samples to document recovery efficiency for the two (2) compounds. Methanol recovery was 89.55% and DCP recovery was 97.13%.

**IV.3 Formaldehyde** - The formaldehyde sampling was performed in accordance with Method 0011. Method 0011 was modified to use midget impingers and sample at a constant rate. Samples were extracted from the exhausts of the Electroless Copper, Semi-Brite Nickel and Brite Nickel Tanks at approximately 1000 cc/per minute through a Teflon sample line and then through midget impingers with 15 mls of DNPH solution in each of the first two (2) impingers. The sampling system used a sampling pump equipped with a calibrated critical orifice. Figure 3 is a schematic diagram of the formaldehyde sampling train.

The samples were analyzed by gas chromatography with a flame ionization detector (GC-FID) for formaldehyde. All the applicable quality assurance and quality control procedures listed in the method were incorporated in the sampling and analysis. In addition, a spiked duplicate train was run during one of the samples to document recovery efficiency for formaldehyde. Formaldehyde recovery was 86.00% on 4/19/17, 84.98% on 4/20/17 and 85.17% on 4/21/17.

**IV.4 Sodium Hydroxide** - The Sodium Hydroxide determinations were performed using a modified version of Method 308. NaOH was captured in deionized/distilled water and analyzed by ion chromatography. Teflon probes were used to extract the exhaust gas from the exhaust. Deionized/distilled water was used to collect the samples. The sampling trains were operated with vacuum pumps with calibrated critical orifices. The orifices were calibrated at approximately 1000 cc/min, Three sixty (60) minute samples were collected from the exhausts. Figure 4 is a schematic diagram of the nitric acid sampling train.

The samples were recovered and refrigerated until they were analyzed. All quality assurance and quality control requirements specified in the method were incorporated in the sampling and analysis. The Sodium Hydroxide recovery was 98.53% on 4/19/17 and 95.33% on 4/20/17.

**IV.5 Total Chrome** - The Cr emission sampling was conducted in accordance with U.S. EPA Method 306. Three (3) samples, 120 minutes in duration each, were collected from the exhaust. The samples were collected isokinetically in 0.1N Sodium Bicarbonate as outlined in the method.

The samples were recovered and analyzed for total chromium by inductively coupled argon plasma/mass spectrophotometry (ICP/MS). All the quality assurance and quality control procedures listed in the method were incorporated in the sampling and analysis. Figure 5 is a schematic diagram of the total chrome sampling train.

**IV.6 Exhaust Gas Parameters -** The exhaust gas parameters (air flow rate, temperature, moisture, and density) were determined by employing U.S. EPA Reference Methods 1 through 4. All the quality control and quality assurance requirements listed in the methods were incorporated in the sampling and analysis.

This report was prepared by:

agl R. Scott Cargill

Project Manager

This report was reviewed by:

David D. Engelhardt Vice President









