1. TEST RESULTS SUMMARY (TRS)

Table 1-1: FPM Results Summary

			Stack Parameters				
			\mathbb{O}_2	CO ₂	Moisture	Temperature	Row Rate
Site	Date	Run	(%)	(%)	(%)	(F)	(DSCFM)
		1	9.8	9.0	15.4	538	30318
RM 05 Outlet		2	9.8	9.0	15.4	557	30240
M O		3	9.5	10.0	15.1	560	30962
		Average	9.7	9.3	15.3	552	30507
			FPM Emissions				
Site	Date	Run	(lbs/ton gla	ss)		(lbs/hr)	(gr/DSCF)
		1	0.06			0.84	0.0032
RM 05 Outlet		2	0.05			0.74	0.0029
RM Out		3	0.01			0.15	0.0006
		Average	0.04			0.58	0.002
Permit Limit		0.45	0.45 n/a		n/a		

Table 1-2: CTM 013 Results Summary

		Stack Parameters				
		O_2	CO ₂	Moisture	Temperature	How Rate
Date	Run	(%)	(%)	(%)	(F)	(DSCRM)
7/14/2020	1	9.8	9.0	14.2	538	30318
7/14/2020	2	9.8	9.0	12.9	557	30240
7/14/2020	3	9.5	10.0	15.3	560	30962
	Average	9.7	9.3	14.13	552	30507
		Er	Emissions			
			H2SO4			
Date	Run	(lbs/ton glass)	(lbs/hr)	(ppmvd)		
7/14/2020	1	0.02	0.27	0.59		
7/14/2020	2	0.02	0.29	0.62		
7/14/2020	3	0.03	0.40	0.86		
	Average	0.02	0.32	0.7		
Permit I	imit_	n/a	1.6	n/a		

Table 1-3: Production Data Summary

Production Data Summary						
Production Rate Pressure Drop						
Run	Time	Tons/Day	Tons/hr	in. WC		
1	0819 - 0939	353.90	14.75	5.6		
2	1033 - 1152	353.90	14.75	5.6		
3	1249 - 1413	353.90	14.75	5.6		

Table 1-4: Summary of Analytical QA/QC Results

Test Method	Parameter	QA/QC Criteria	Ground Site QA/QC Status	Outlet Site QA/QC Status	Within QC Criteria?
RM 2	Pitot Leak Check	$\Delta~0.0''~\text{H}_2\text{O}~/~15~\text{seconds}$		0.0 @ 4.2" (max)	Yes
RM 5	Sample Train Leak Check (post test)	<0.02 cfm		0.001 cfm @ 5" H ₂ O (max)	Yes
RM5	Isokinetics	100% +/- 10%	ngun dinamia nangananangangun dinambananan Paganangan dinamangan dinamanangan dinambananan Paganangan dinambanangan dinambanangan dinambanangan dinambanan	100.4 - 101.2%	Yes
	Sample Train Leak Check (post test)	<0.02 cfm	0.017 cfm @ 11.0" H ₂ O (max)		Yes
CTM013	Probe Temperature	> 350 °F	358°F (avg.)		Yes
	Thimble Temperature	> 500 °F	510°F (avg.)		Yes



2. INTRODUCTION

2.1 Introduction

Guardian Industries, LLC (Guardian) has contracted Empire Stack Testing, LLC. (Empire) to perform Filterable Particulate Matter (FPM) and Sulfuric Acid (H_2SO_4) testing services on their glass furnace in Carleton, Michigan. Testing used RM5 at the Trimer outlet stack, and CTM-13 at the outlet ground site of the Trimer control system.

Section 3 of this protocol contains the sampling and analytical procedures used to perform the test program. Section 4 details the quality assurance/quality control (QA/QC) procedures for the test program.

2.2 Test Program Objective

This test program is required annually to quantify the FPM and H_2SO_4 emissions from the outlet of the Trimer control system. All testing followed applicable methodologies of the Environmental Protection Agency (EPA), and as defined in Table 3-1, below.

2.3 Test Personnel

Coordinating the test program were: Benjamin Kroeger, CSP Guardian Industries, LLC. (734)-654-4430

Ancy Sebastian ALS Environmental (905)-331-3111

Michael T. Karter, QSTI Empire Stack Testing, LLC. (716)-481-6749

2.4 Test Plan

Testing for all parameters was completed in triplicate following Reference Methods (RMs). The test program incorporates reference methods outlined in the United States Environmental Protection Agency (USEPA) Code of Federal Regulations Title 40, Part 60 (40CFR60), Appendix A. See Table 2-1 below.

2.5 Test Schedule

Day 1 (7/13/2020): Mobilize to Guardian and finish setup for testing Day 2 (7/14/2020): Complete FPM & H2SO4 Testing (\sim 8 hours)

Day 3 (7/15/2020): Demobilize from site

Table 2-1: Summary of Test Plan

PARAMETER	METHOD	ANALYSIS	SAMPLE DURATION (MINUTES)	TEST LOCATION(S)
Flow Rate	RM 1 & 2	S-Type Pitot Tube / Manometer	various	Outlet
Dry Molecular Weight	RM 3	O ₂ and CO ₂ Fyrites	various	Outlet & Outlet GS
Moisture	RM 4	Gravimetric	30	Outlet & Outlet GS
FPM	RM 5	Gravimetric	60	Outlet
H ₂ SO ₄	CTM 013	Titration	30	Outlet GS

NOTES:

CTM: FPM: Conditional Test Method Filterable Particulate Matter

GS:

Ground Site

H₂SO₄: RM: Sulfuric Acid United States Environmental Protection Agency Reference Method

2.6 Process Description

Flat glass manufacturing Line #1 consisting of a raw material melting furnace, glass forming and finishing, and glass cutting. Line #1 produces flat glass using the float method. Materials are weighed and mixed with water in the batch house before entering the natural gas fired Furnace. Glass then enters the tin bath to be formed and drawn. Next, it enters a lehr to reduce its temperature. The emission unit is controlled by a new (Trimer ECS) Control Device consisting of a Dry Scrubber, Particulate Filter, and Selective Catalytic Reduction (SCR).

2.7 Plant data

The plant's SCADA system continuously records the operating data included in the test report. The plant shall provide plant operation and summarize pertinent operating data to represent plant operation. These data and summaries are provided both electronically (MS Excel) and in paper copies.

3. PRESENTATION OF RESULTS / EXECUTIVE SUMMARY

This Executive Summary discusses, in detail, the test results and any anomalies, their resolution, and any effect on the results quality or usability.

3.1 Discussion of Results

Testing was completed on July 14th, 2020 for FPM, H₂SO₄, and SO₂. During this test program, the facility operated at an average production rate of 353.9 tpd.

The results indicate that the measured emissions are compliant with their permit limits. All field and lab data are included in the appendices of this report.

Isokinetics 3.1.1

Each RM 5 sample run for FPM met the isokinetic limit of 100 % \pm 10%. These and other QAQC criteria are summarized in Table 1-4.

3.1.2 FPM Test Result (RM 5)

The average FPM emissions were measured to be 0.04 lbs/ton; which is compliant with limit of 0.45 lbs/ton. See Summary Table 1-1.

3.1.3 H₂SO4 Test Result (CTM 013)

The average emission rate of sulfuric acid was 0.32 lbs/hr and 0.02 lbs/ton of glass. The unit demonstrated compliance with the emission limit of 1.6 lbs/hr. See Table 1-2.

Anomalies 3.2

No anomalies were recorded during testing, analysis, nor report production.

·				

4. SAMPLING AND ANALYTICAL PROCEDURES

This section provides a brief overview of the specific test methods that were used to determine the Sulfuric Acid emissions from each the glass furnace. All test method procedures were performed in accordance with the USEPA Reference Methods given in 40CFR60, Appendix A. The details of each method are given in the following sections.

4.1 **Reference Method Test Location**

The exhaust stack is fixed with four 10-inch diameter ports. The test ports are located approximately 7 equivalent diameters downstream of a disturbance and 11 equivalent diameters upstream of another disturbance. See Figure 4-1.

4.2 **Sampling Point Location**

Volumetric Flow 4.2.1

Representative measurement of pollutant emissions and total volumetric flow rate from a stationary source requires a measurement site where the effluent stream is flowing in a known direction and cyclonic flow is not present. See section 4.3.1, below.

According to Reference Method 1, the cross section of the stack is divided into equal areas and a traverse point is then located within each of these areas. The number of duct diameters upstream and downstream from the test location to a flow disturbance determines the number of traverse points in a cross section.

As these stacks have diameters >24 inches the outermost traverse points were at least 1 inch from the stack walls.

Sampling was performed at six traverse points per each of the four test ports for a total of 24 sampling points, as set forth by RM 1. See Figures 3-3 and 3-4.

4.3 Stack Gas Velocity and Volumetric Flow Rate

According to Reference Method 2, the gas velocity in a stack were determined from the average velocity head with a type S Pitot tube, gas density, stack temperature, and stack pressure.

The average velocity head were determined by using an inclined manometer and a type S Pitot tube with a known coefficient of 0.84 that is determined geometrically by standards set forth in Reference Method 2. Stack temperature were taken at each traverse point using a type K thermocouple. Static pressure was determined by using a straight tap and an inclined manometer.

4.3.1 Cyclonic Flow Check

The cyclonic flow check was performed during prior to testing and demonstrated non-cyclonic, laminar flow. These data are included in the test report.

4.4 Oxygen & Carbon Dioxide Concentration (RM 3)

Oxygen and Carbon Dioxide concentrations were quantified on a dry basis according to RM-3A. Stack gas was analyzed continuously with a Servomex Model 1400 paramagnetic O_2 analyzer and a CAI Model 10 NDIR CO_2 analyzer, or equivalent. See Figure 3-5.

The corrected gas concentrations were calculated using equation 7E-5b below referenced in 40CFR60, App. A, RM-7E, Section 12:

$$C_{G_W} = (C_{Avg} - C_O) \frac{C_{MA}}{C_{M} - C_O}$$
 Eq. 7E-5b

Where:

C_{gas} = Effluent gas concentration, dry basis, ppm.

C_{avg} = Average gas concentration indicated by gas analyzer, dry basis, ppm.

C_o = Average of initial and final calibration bias check responses for the zero gas, ppm. = Average of initial and final system calibration bias check responses for the upscale

calibration gas, ppm.

C_{ma}= Actual concentration of the upscale calibration gas, ppm

4.5 Moisture Determination (RM 4)

The determination of effluent moisture was performed as part of the wet-chemistry sampling, as detailed below in RM 5 and CTM013.

4.6 Filterable Particulate Matter (RM 5)

4.6.1 Background

Reference Method 5 were used to determine the FPM concentrations. An integrated sample were drawn from the stack. The filterable particulate was quantified from the probe and filter catch.

4.6.2 Sampling

An isokinetic sample were collected at a rate of approximately 0.7 cubic feet per minute (cfm) for 72 minutes. A heated glass probe, heated glass filter, and standard full-size impingers were used. The first two impingers each contained 100 ml each of distilled water. The third impinger remained empty. The last impinger contained a known

amount of silica gel. The second impinger is a Greenburg-Smith design; the remaining impingers are modified Greenburg-Smith designed. A schematic of the sampling train is presented in Figure 4-7. Both the probe and filter were maintained at 250 °F, ±50 °F as required by the method.

Sample Recovery 4.6.3

Recovery of all sample train components was performed in Empire's Mobile Laboratory.

Container 1:

The filter was carefully removed from the filter holder with the use of tweezers and disposable surgical gloves, and placed into its Petri dish labeled with the filter ID number and identified as "Container No. 1" for the proper run and location. Any particulate matter and/or fiber filters that adhered to the filter holder or filter holder gasket were carefully transferred to the Petri dish with the use of a dry nylon bristle brush or a sharp-edged blade. The Petri dish were then sealed with parafilm. The probe nozzle, probe liner, and front half of the filter holder were rinsed at least three times with acetone, and the rinses collected in a sample jar labeled "Container No. 2". The container was then sealed, and the fluid level marked.

Container 2:

The particulate matter was recovered from the probe nozzle, union, probe liner, front half of the filter holder, and (if applicable) the cyclone, as follows;

- a. The nozzle was rinsed with acetone, brushed with a non-metallic bristle brush, and rinsed with acetone until no visible particles remained. A final acetone rinse was performed.
- b. The probe liner was rinsed and brushed at least three times, followed by a final rinse of the brush with acetone.
- c. After completing the rinses, the lid on the sample container were tightened and the height of the fluid level marked.

Acetone Blank:

An acetone blank with a volume roughly equal to the rinse volume were saved as a blank.

4.6.4 **Analysis**

The samples were shipped to ALS Global (ALS) for analysis following RM 5. The filters are desiccated to a constant weight. The gravimetric analysis of the filters and acetone samples were repeated every six to twenty-four hours until stable analyses are obtained.

ALS uses a 40 mL vial to analyze the acetone rinses, in lieu of evaporation in a 250 mL beaker. This minimizes the tare weight of the vessel; as the vials have a tare weight of

approximately 21g compared to a tare weight of approximately 100g with a 250 mL glass beaker. The 250 mL glass beaker has a greater chance of variability; also, the NJ-DEP (the primary NELAC accreditor) has certified ALS to perform this analysis with the modification listed.

The procedure used is as follows:

- The vials are kept in the balance room at all times prior to use. Lab numbers are put on the vials with a black magic marker and the vial is then desiccated for one hour prior to doing the pre-weight
- Place bottle of solvent onto Navigator balance, enter the weight into the "Bottle and Solvent Weight" column
- Place a ribbed watch glass on the sample container and set in a fume to evaporate to <10 mL
- Transfer the remaining solvent to a pre-cleaned, pre-weighed and pre-numbered 40 mL glass vial
- Place the empty bottle of solvent onto Navigator balance, enter weight into the "Empty Bottle Weight" column
- Reduce to dryness with a gentle stream of N2 using the N-Evap system
- Place vials in desiccators for 24 hours minimum and record the time in the spreadsheet
- Note the appearance of the residue on the worksheet, (light, dark, minimal, copious as I/d/m/c)
- Proceed to 7.4 (Balance use and weighing samples)
- When all weightings are complete a second analyst must select and reweigh 1 of every 10 vials (the vial is to be selected at random) Second analyst's result must be ± 2 mg of the first analyst's result.

4.7 **Sulfuric Acid (CTM-013)**

4.7.1 **Background**

This method was developed as an alternative to EPA Method 8 for determining sulfuric acid emissions from Kraft recovery furnaces. When testing recovery furnaces, EPA Method 8 is subject to significant interference from sulfates, which are present in the particulate matter, and sulfur dioxide. The alternative method uses a quartz in-line thimble to remove particulate matter from the gas stream prior to capturing sulfuric acid. The use of a controlled condensation technique eliminates the potential for interference from sulfur dioxide.

A gas sample is extracted from the sampling point in the recovery furnace stack. The sulfuric acid vapor or mist (including sulfur trioxide) and the sulfur dioxide are separated, and both fractions are measured separately via titration.

4.7.2 Sampling

The sampling train consists of a glass nozzle and heated glass probe, which were maintained at the temperature of >177°C (350°F). The probe was then connected to the thimble holder housed in an oven box that were also maintained at the temperature of >500 °F. The thimble holder was constructed of quartz with a quartz thimble filter.

Sampling were performed for a minimum of 30 minutes at a constant rate ($\pm 10\%$) of ~10.0 lpm (~0.35 cfm).

A condenser connects the thimble to the train. The condenser is filled with water and its temperature is maintained between 75 and 85°C (167 to 185°F). The condenser was connected to the impinger train with a minimal length of unheated Teflon tubing. The first and third impingers consist of Greenburg-Smith design, the remaining impingers are modified Greenburg-Smith designed impingers. The first two impingers contained 100 ml of 3% hydrogen peroxide (H₂O₂). The third impinger contained 100 ml of distilled deionized water (RODI). The fourth impinger contained approximately 500 g of silica gel desiccant.

A vacuum line connects the outlet of the last impinger to the control module. The control module consists of a vacuum gauge, rotary pump, by-pass and main valve, dry gas meter, orifice, and an inclined manometer. The sample train is illustrated in Figure 4-8.

Coinciding with the sampling were velocity, moisture, and dry molecular weight determinations.

Sample Purge 4.7.3

At the completion of the test run, the probe was separated from the thimble, and a 15minute purge with clean air (ambient) were performed at the same rate at the test run, as required by the method.

Sample Recovery 4.7.4

Recovery were performed onsite in Empire's mobile laboratory at the completion of each test run.

Container 1:

Rinse separately the probe, quartz thimble holder and the H₂SO₄ condenser with deionized water using multiple rinse. After completing the rinses, the lid on the sample container were tightened and the height of the fluid level marked. The thimble was discarded.

Guardian	Industries	Corp.
on the Tri	mer Control	System

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Container 2:

The liquid from the first two impingers were quantitatively transferred into a clean sample bottle (glass or plastic). (SO₂ analysis only)

Container 3:

The water from the third impinger were weighed in the field, and then discarded.

Blank H₂O₂:

Take ~ 100 ml of H_2O_2 and place it in a recovery bottle. The liquid level on the bottle were marked.

Blank H₂O:

Take $\sim \! 100$ ml of H_2O and place it in a recovery bottle. The liquid level on the bottle were marked.

4.7.5 Analysis

The **container-1 and blank** samples were shipped to ALS Environmental of Mississauga, Ontario, Canada for H_2SO_4 analysis **via titration**.

Figure 4-1: Test Port Location (Outlet)

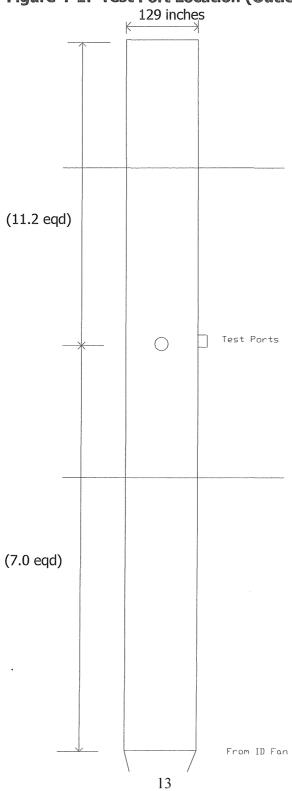




Figure 4-2: Sampling Point Locations (Outlet)

Traverse Point Number	Distance from Inner Wall (%)	Distance from Port Edge (inches)
1	2.1	48.7
2	6.7	54.6
3	11.8	61.2
4	17.7	68.8
5	25.0	78.3
6	35.6	91.9
7	64.4	129.1
8	75.0	142.8
9	82.3	152.2
10	88.2	159.8
11	93.3	166.4
12	97.9	172.3
Diameter: 129"		

46"

Nipple:

Figure 4-3: RM 5 Sampling Train

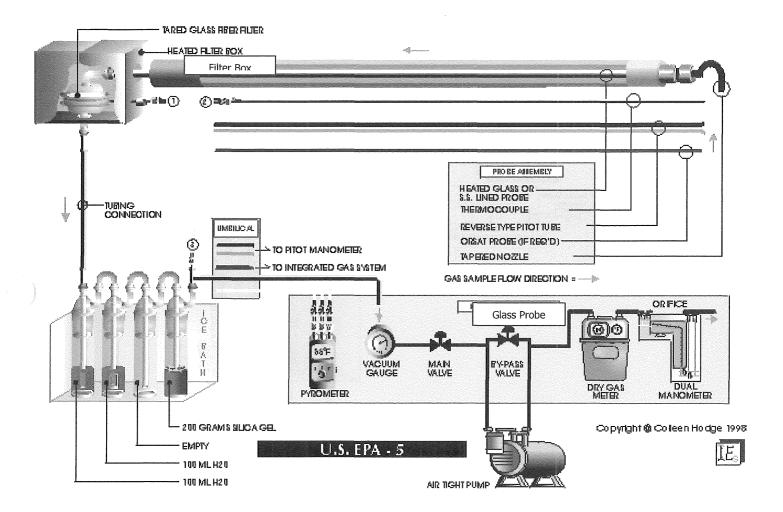
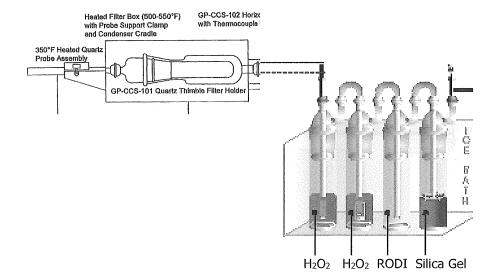


Figure 4-4: CTM 013 Sampling Train



5. QUALITY ASSURANCE AND QUALITY CONTROL (QA/QC)

Quality control procedures for all aspects of field sampling, sample preservation and holding time, reagent quality, analytical methods, analyst training and safety, instrument cleaning, calibration, and safety were followed. These procedures were consistent with EPA Guidelines documented in:

EPA 600/9-76-005, Quality assurance Handbook for Air Pollution Measurement Systems, Volume I EPA 454/R-98-004, Quality assurance Handbook for Air Pollution Measurement Systems, Volume II EPA 600/R-94-038c, Quality assurance Handbook for Air Pollution Measurement Systems, Volume III

5.1 **Chain of Custody**

Documentation of the Chain-of-Custody of samples and data obtained during the test program is essential for ensuring the validity of the test program results. Chain-of-Custody procedures were followed during sampling, sample and data transport, sample preparation and analysis, storage of data, as well as with archived samples and reported results. Empire follows the protocol listed in SW 846, Section 1.3 during field sampling and in-house laboratory analysis.

5.2 **Equipment and Sampling Preparation**

Sampling equipment were cleaned, checked, and calibrated prior to use in the field. Each parameter's sampling method requires specific cleaning methods of the glassware, train components, and recovery containers. These materials were then sealed prior to shipment to the field.

5.3 **Calibrations**

Pitot Calibration 5.3.1

Pitot tubes were calibrated according to Reference Method 2, Section 10.1. Pitot tubes were given a baseline coefficient of 0.84 when they meet certain geometrically measured angles and dimensions as set forth in the method.

Thermocouple Display Calibration 5.3.2

Following Method 2, Section 10.3, an NIST Traceable Electronic Thermocouple Calibrator/Simulator (ALTEK) for post-test calibrations is used. If the display being calibrated and the ALTEK were within +/-1°F and/or +/-2% of the reference temperature, the calibration is acceptable, else the display is re-calibrated.

5.3.3 Thermocouple Calibration

According to EMTIC GD-28, a single point (at ambient temperature) check of the thermocouple were made prior to and following each test program. If the thermocouple being calibrated and the certified thermometer were within \pm 0 °F of each other, the calibration is acceptable. The thermocouple must also respond appropriately to a change in temperature. Thermocouples that fail either of these criteria were repaired or discarded.

5.3.4 Barometer Calibration

During testing, the barometric station pressure was obtained online from the nearest NOAA or FAA weather station.

5.4 Leak Checks

5.4.1 Sample Trains (CTM013)

A leak-check prior to the sample run is optional; however, a leak-check after the sampling run is mandatory. The leak check was conducted in accordance with the procedures outlined in Reference Method 5, Section 8.5.9, except that it was conducted at a vacuum equal to or greater than the maximum value reached during the sampling run. If the leakage rate is found to be no greater than 0.02 cfm, the results were acceptable, and no correction were applied to the total volume of dry gas metered.

5.4.2 Sample Trains (FPM)

Both pre- and post-run leak checks were conducted. A pre-test leak check was performed to verify integrity of the vacuum system. A leak check is mandatory at the conclusion of each isokinetic sampling run. The leak check was conducted in accordance with the procedures outlined in Reference Method 5, Section 8.5.9, except that it was conducted at a vacuum equal to or greater than the maximum value reached during the sampling run. If the leakage rate is found to be no greater than 0.02 cfm, the results were acceptable, and no correction were applied to the total volume of dry gas metered.

5.4.3 Pitot Leak Check

The pitot tubes used during the test program were leak checked prior to the test series and following each traverse set, as prescribed in RM 2, Section 8.1. The leak check was performed by pressurizing the positive side of the pitot to at least 3 inches of water. No loss of pressure for 15 seconds indicates a successful leak check. This procedure was repeated with a vacuum applied to the negative side of the Pitot tube as well.

CEMS Sampling Performance Specifications 5.5

5.5.1 **System Response Time**

The system response time was initially checked during the site set-up activities according to Method 7E. These results are included in Appendix A of this report.

Analyzer Calibration Error 5.5.2

The purpose of the Analyzer Calibration Error (ACE) procedure was to establish an initial calibration curve and to assure that each calibration point was accurate to within 2% of the calibration span (CS) value. This was accomplished following the procedures outlined in Methods 3A and 7E by first adjusting the monitors to the low and CS calibration gases. Then the low-level, mid-level and CS gases were injected and their responses noted to verify instrument linearity.

5.5.2.1 **Oxygen**

The RM O_2 analyzer was operated in the 0-25% (volume) range. EPA Protocol calibration gases were used at concentrations equivalent to approximately 0-20%, 40-60%, and the CS concentration.

5.5.2.2 Carbon Dioxide

The RM CO₂ analyzer was operated in the 0-25% (volume) range. EPA Protocol calibration gases were used at concentrations equivalent to approximately 0-20%, 40-60%, and the CS concentration.

5.5.3 **System Bias Check**

The sample system bias (SB) as required in EPA Methods 3A was determined. Following the ACE, the calibration gases are injected into the sample manifold and in turn through the entire sample path. The system responses are noted and compared to the CE values. If the two calibrations differ by less than 5% of the CS, the check was considered valid.

5.5.4 **Drift Assessment**

A drift assessment was performed following the procedures outlined in Method 7E. Immediately following each test run, the low-level gas was introduced into the system and the monitor's response recorded. If the response does not vary (drift from) from the previous value by more than 3% of the CS, the assessment was considered valid. This procedure was repeated for the upscale calibration gas.

5.5.5 Interference Response

Interference response data was available for all gas analyzers used during this program. This data shows that each analyzer meets the interference criteria of the methods. An interference response has been determined using the procedure in RM 7E, Section 8.2.7. Any interference detected was within the allowable limits.

5.6 Sample Recovery

All sample volumes and reagent volumes were measured and recorded on Empire's recovery data sheets and included in the report. All recovery procedures were intended to meet the requirements of the methods.

5.7 Data Reduction

The QA/QC procedures for data reduction include using computer programs to generate tables of results. Results for at least one test run were double-checked and recalculated by hand. These pages are included in the report.

The wet-chemistry data were logged directly to a separate laptop hard drive, where calculations were performed using MS-Excel spreadsheets. These data were archived nightly to flash media. Copies of these data were available in the field electronically or in print form, upon request. Paper datasheets are only used in an emergency and were not used during this test program.

5.8 Safety

These methods involve hazardous materials, operations, and equipment. Empire established appropriate safety and health practices and determined the applicability of regulatory limitations before performing this test program.

The test site shall meet the criteria of RM 1. Test ports (loosened and cleaned), safe access, and suitable power to be provided by the client. The above items need to be ready upon arrival of the test crew.

Delay or Lost Time (delays) of the field crew due to causes beyond the control of Empire Stack Testing, LLC. (Empire) may include (but were not limited to weather, cyclonic flow conditions, process upsets or failure, or the facility's inability to maintain the desired test conditions). Inclement weather includes (but is not limited to) lightning, strong rains, blizzards, high winds (\geq 30 mph), high humidity, and/or working temperatures below 20 °F or above 90 °F. Empire's field leader retains the right of final refusal to stop testing for any unsafe condition.