

MICHIGAN DEPARTMENT OF ENVIRONMENT, GREAT LAKES, AND ENERGY

INTEROFFICE COMMUNICATION

TO: File for Propyl Alcohol (CAS # 71-23-8)

FROM: Keisha Williams, Air Quality Division

DATE: May 8, 2019

SUBJECT: Screening Level Update for Propyl Alcohol

The initial threshold screening level (ITSL) for propyl alcohol is 2500 $\mu\text{g}/\text{m}^3$ (8-hour averaging time) based on the Michigan Department of Environment, Great Lakes, and Energy* (EGLE), Air Quality Division (AQD) Rule 336.1232 (1)(c) and (2)(a). The chronic ITSL of 730 $\mu\text{g}/\text{m}^3$ (annual averaging time) is being rescinded at this time.

The following references or databases were searched to identify data to determine the screening level: United States Environmental Protection Agency's (EPA's) Integrated Risk Information System (IRIS), ChemView: the EPA's database on chemical health and safety data for chemicals subject to the Toxic Substances Control Act (TSCA), the TSCA documents in the National Technical Reports Library (NTRL) database, the American Conference of Governmental Industrial Hygienists (ACGIH) Threshold Limit Values (TLV), National Institute of Occupational Safety and Health (NIOSH) Pocket Guide to Hazardous Chemicals, Agency for Toxic Substances and Disease Registry (ATSDR) Minimal Risk Levels, International Agency for Research on Cancer (IARC) Monographs, the American Chemical Society's SciFinder database, Health Effects Assessment Summary Tables (HEAST), National Toxicology Program (NTP) Study Database, EPA Superfund Provisional Peer Reviewed Toxicity Values, EPA Acute Exposure Guideline Levels (AEGs) for Airborne Chemicals, United States Department of Labor Occupational Safety and Health Administration (OSHA) Permissible Exposure Limits (PELs), the Canadian Centre for Occupational Health and Safety's Registry of Toxic Effects of Chemical Substances (RTECS), the Toxnet databases: Hazardous Substances Data Bank and Toxline, Spacecraft Maximum Allowable Concentrations (SMACs), California Office of Environmental Health Hazard Assessments Reference Exposure Levels, Texas Commission on Environmental Quality (TCEQ) Effects Screening Levels (ESLs), German Maximale Arbeitsplatz-Konzentration (MAK) values, and European Chemicals Agency Registered Substances Dossiers.

Background Information

Propyl alcohol, also known as n-propanol and 1-propanol, has been used as a solvent for several products including foods, inks, and pesticides (ACGIH, 2007; HSDB, 2001). It has also been used as an antiseptic, degreasing agent, and disinfectant. The chemical structure is provided in Figure 1 and chemical properties are listed in Table 1.

*Formerly the Michigan Department of Environmental Quality (MDEQ).

Figure 1. Chemical structure for propyl alcohol

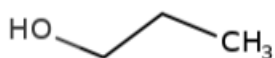


Table 1. Chemical and physical properties of propyl alcohol

Molecular weight: 60.096 grams/mole
Melting point: -127 °C
Boiling point: 97.2 °C
Vapor pressure: 21.0 mmHg at 25°C
Henry's Law Constant: 7.41e-06 atm-m ³ /mole
Vapor density: 2.1, where air=1
Physical state at standard temperature and pressure: liquid
Odor: mild, alcohol-like odor
Odor perception threshold: 0.07-100 mg/m ³

Reference: PubChem Compound Database

There are noted deficiencies in the toxicity database of propyl alcohol, especially the lack of chronic or subchronic studies (EPA, 1997; EPA, 2007; ECB, 2008). However, propyl alcohol is expected to have similar toxic effects as alcohols that have more extensive toxicity databases: isopropyl alcohol and ethanol. In general, inhalation of these toxic air contaminants may cause irritation, neurotoxicity, and developmental/reproductive toxicity (HSDB, 2001). Furthermore, these alcohols are all readily metabolized by alcohol dehydrogenase. However, subsequent metabolites, relative toxicity and respective critical effects can differ among these alcohols (Nelson et al., 1988; ACGIH, 2001; ACGIH, 2007; ACGIH, 2008; MDEQ, 2016, MDEQ, 2017). As a result, isopropyl alcohol and ethanol are described here for hazard identification.

Occupational exposure limits (OELs) are the main inhalation health benchmarks that have been derived for propyl alcohol specifically (ACGIH, 2007; NIOSH Pocket Guide; OSHA Chemical Database) (Table 2). The critical effects cited are narcosis and/or irritation (ACGIH, 2007; NIOSH Pocket Guide, OSHA Chemical Database). The health effects screening levels designed by the TCEQ are based on the ACGIH OELs (TCEQ, 2015). The EPA has determined that there is not enough information for derivation of a subchronic or chronic reference concentration (RfC) (EPA, 1997; EPA, 2007).

Table 2. Occupational Exposure Limits for Propyl Alcohol

Agency Reference	Benchmark Value for Agency-Specific Time Weighted Average OEL
NIOSH	Reference effect levels (RELs)*: 500 mg/m ³
ACGIH	Threshold limit value (TLV): 246 mg/m ³
OSHA	Permissible exposure limit: 500 mg/m ³

* NOTE: Was also given a short-term exposure limit, 625 mg/m³ and a skin notation

As compared to the OELs that were designed to be health-protective of narcosis and/or irritation, the developmental study in rats by Nelson et al., 1988 has been used to derive EGLE screening levels (EGLE Chemical Criteria Database), where the developmental effects of fetal skeletal malformations and decreased fetal body weight gain were observed.

Key studies reviewed for ITSL derivation

Because propyl alcohol is expected to be absorbed into the blood following inhalation exposure, portal of entry effects may not be the critical effects, and there are longer duration studies that have been conducted via oral administration as compared to inhalation exposure, the oral toxicity data were assessed. Risk assessments published by both the EPA and European Chemicals Bureau (ECB) in 2007 and 2008, respectively identified nearly the same oral, repeated dose studies where hepatotoxicity was identified as the critical effect (EPA, 2007; ECB, 2008). With this review, no other relevant oral studies were identified following the 2007 and 2008 evaluations.

Studies by Gibel et al. were identified as key studies to discuss here, because they are the only propyl alcohol-specific lifetime rodent studies (Gibel et al, 1974; Gibel et al., 1975). Furthermore, the Gibel et al. studies presented the lowest observable adverse effect level (LOAEL) identified out of the oral studies presented in the 2007 and 2008 evaluations. However, it is important to note that both the EPA and ECB concluded that the oral toxicity database was insufficient for health benchmark derivation. For the purposes of this evaluation, a potential ITSL based on the lifetime oral studies conducted by Gibel et al. will be further described.

These studies were published in German, but a translation is available through the National Technical Reports Library (EPA, 1992). In the Gibel et al. studies, 0.3 mL/kg propyl alcohol was given twice a week by gavage for a lifetime in male and female Wistar rats (EPA, 1992; EPA, 2007). Using a specific gravity of 0.804 grams/mL, this dose was estimated to be 241 mg/kg (EPA, 2007). Time matched controls were given a solution of 0.9% sodium chloride. While no malignant tumors were observed in the control group (N=25), 5 malignant tumors were observed in rats given propyl alcohol (N=18). More specifically, the 5 malignant tumors were “two myeloid leukemias, one liver-cell carcinoma, two liver sarcomas” (EPA, 1992). Furthermore, while 3 benign tumors were observed in the control group, 10 benign tumors were observed in the propyl alcohol group. It is unclear whether these tumors occurred in different animals or simultaneously in the same animal. Besides carcinogenicity, the Gibel et al. study reports, “...we must emphasize the severe toxic damage to the liver observed in almost all the animals, regardless of the nature of the alcohol, and also the hyperplasia of the hematopoietic parenchyma...Of n-propyl alcohol, we can speak of a hepato and hemotoxic effect, as well as a carcinogenic one” (EPA, 1992). Taking the administered dose as the LOAEL for hepatotoxicity and hemopoietic toxicity, a chronic ITSL can be derived as shown in Equation 1 pursuant to AQD Rule 336.1232 (1)(b) and (2)(b).

Equation 1.

$$ITSL = RfD \times \frac{70kg}{20 m^3}$$
$$RfD = \frac{HED}{UFs}$$
$$HED = POD_{adj} \times \frac{animal\ BW^{-0.25}}{human\ BW^{-0.25}}$$
$$POD_{adj} = LOAEL \times \frac{dose\ per\ week}{7\ days\ per\ week}$$

Where:

- POD_{adj} =time adjusted point of departure
- $LOAEL=241$ mg/kg
- Body weight of rat=0.3795 kg (MDEQ, 1996)
- Body weight of a person=70 kg
- UFs=uncertainty factors=10 for LOAEL to NOAEL extrapolation, 3 for interspecies extrapolation and 10 for intraspecies extrapolation

$$POD_{adj} = 241 \frac{mg}{kg} \text{ per day} \times \frac{2 \text{ doses given per week}}{7 \text{ days per week}} = 68.85714286 \frac{mg}{kg} \text{ per day}$$

$$HED = 68.85714286 \frac{mg}{kg} \times \left(\frac{(0.3795 \text{ kg})^{-.25}}{(70 \text{ kg})^{-.25}} \right) = 250.5996601 \frac{mg}{kg} \text{ per day}$$

$$RfD = \frac{250.5996601 \frac{mg}{kg} \text{ per day}}{10 \times 10 \times 3} = 0.8353322 \times \frac{mg}{kg} \text{ per day}$$

$$\begin{aligned} \text{Potential ITSL} &= 0.8353322 \times \frac{mg}{kg} \text{ per day} \times \frac{70 \text{ kg}}{20 \text{ m}^3} \times \frac{10^3 (\mu g)}{mg} = 2923.662701 \frac{\mu g}{\text{m}^3} \\ &\approx 2900 \frac{\mu g}{\text{m}^3}, \text{ annual averaging time} \end{aligned}$$

The Gibel et al. studies were the only studies found that looked at the carcinogenic effects of propyl alcohol over a lifetime of oral administration (EPA, 1992; EPA, 2007; ACGIH, 2007). However, there are major limitations with the study that prevent the use of these studies for initial risk screening level (IRSL) derivation, specifically only one treatment dose was used and the description of quantification of tumors in the individual rats in the dose group was not clearly presented. So, while EPA has classified propyl alcohol as a possible human carcinogen based on limited animal data, a cancer slope factor has not been derived. As a result, AQD will not regulate propyl alcohol with an IRSL at this time.

The ITSL established in 1992 was based on an inhalation study, where pregnant rats were exposed to 0, 3500, 7000 or 10000 ppm for 7 hours per day for 20 days during gestation (MDNR, 1992 [attached]; Nelson et al., 1988). The ITSL justification document further states, "Fetal skeletal malformations were increased, and fetal body weights were decreased in fetuses from dams exposed to 7000 and 10000 ppm. From this study a no observable adverse effect level (NOAEL) of 3500 ppm (or 8750 mg/m³) can be identified". The ITSL was calculated as shown in Equation 1 pursuant to AQD Rule 336.1232 (1)(d).

Equation 1.

$$ITSL_{\text{established in 1992}} = \frac{NOAEL}{35 \times 100} \times \frac{\text{hours exposed}}{24 \text{ hours per day}}$$

Where:

- NOAEL is 8750 mg/m³
- Hours exposed were 7 hours per day
- Uncertainty factors (UFs) were 35 for 7 day duration study extrapolation to chronic duration, 10 for interspecies extrapolation, and 10 for intraspecies extrapolation

$$ITSL_{established\ in\ 1992} = \frac{8750 \frac{mg}{m^3}}{35 \times 100} \times \frac{7}{24} = 730 \frac{\mu g}{m^3}, \text{ annual averaging time}$$

Because the critical effect is a developmental effect, it is more appropriate to follow EPA guidance and recommendations on RfC derivation with consideration for developmental effects (EPA, 1991; EPA, 2002; EPA, 2012) as shown in Equation 2.

Equation 2.

$$ITSL = RfC_{based\ on\ developmental\ effects} = \frac{HEC}{UFs}$$

$$HEC = POD_{adj} \times DAF$$

$$POD_{adj} = POD \times \frac{\text{hours of exposure}}{7 \text{ hours}} = 8750 \frac{mg}{m^3} \times \frac{7}{24} = 2552.083 \frac{mg}{m^3}$$

Where:

- POD_{adj} is the time adjusted point of departure, where the NOAEL from the Nelson et al. (1988) study is used as the point of departure (POD).
- DAF= dosimetric adjustment factor for category 2 gas with systemic effects= default value=1
- UFs=uncertainty factors of 3 for interspecies differences and 10 for intraspecies differences

$$HEC = 2552.083 \frac{mg}{m^3} \times 1 = 2552.083 \frac{mg}{m^3}$$

$$Potential\ ITSL = \frac{2552.083 \frac{mg}{m^3}}{3 \times 10} \times \frac{10^3 \mu g}{mg} = 85,069.43 \frac{\mu g}{m^3} \approx 85,000 \frac{\mu g}{m^3}, 24 \text{ hour averaging time}$$

A summary of an inhalation study referenced from a 1992 report is available in the ECHA dossier for propyl alcohol (ECHA, 2018). In this study, groups of male and female rats were exposed to either 0, 100, 500 or 1000 ppm propyl alcohol for 6 hours per day for a total of 9 exposures on consecutive weekdays. The sample size was 15 rats per sex for the 0 and 1000 ppm groups, and 10 rats per sex for the 100 and 500 ppm groups. It was noted that perinasal and periorcular encrustation was observed in 1 male in the group exposed to 500 ppm propyl alcohol. In the summary conclusion, it further notes “9 days of repeated inhalation exposures to n-propyl alcohol vapor produced only minimal clinical signs, most of which were observed only at the highest exposure concentration of 1000 ppm... Exposure-related clinical observations were limited to swollen periorcular tissue, and perinasal and periorcular encrustation in the 1000 ppm exposure concentration group.” The summary states that “no treatment-related systemic effects” were observed as measured by organ weight changes, body weight changes, urinalysis, hematology, evaluation for gross lesions during necropsy, and histopathology.

This study summary shows point of contact effects at concentrations below the NOAEL observed in the Nelson et al. studies. However, it is important to note that the clinical observations described above are characterized in the ECHA summary as the lowest observable effect concentration (LOEC), and not the LOAEL. The summary identified no critical effects, indicating that the local effects were determined to not be classifiable as “adverse.”

However, the local effects are dose-dependent effects that suggest point of contact effects are the critical effects if a high enough concentration is reached. Using 500 ppm as a NOAEL, potential ITSLs can be derived as shown in Equations 3 and 4 pursuant to AQD Rules 336.1232 (1) (d) and 336.1233 (1), respectively.

Equation 3.

$$Potential\ ITSL = \frac{NOAEL}{35 \times 100} \times \frac{hours\ exposed\ per\ day}{24\ hours\ per\ day}$$

Where

- NOAEL=500 ppm or 500ppm x 0.0409 x 60.096 grams/mole=1228.9632 mg/m³
- Hours exposed =6 hours per day
- UF of 20 is used instead of 35 for chronic extrapolation because the study is longer than 7 days.

$$Potential\ ITSL = \frac{1228.9632 \frac{mg}{m^3}}{20 \times 100} \times \frac{6\ hours}{24\ hours\ per\ day} \times \frac{9\ days}{11\ days} \times 1000 \frac{\mu g}{mg} = 125.6894182$$

$$\approx 126 \frac{\mu g}{m^3},\ annual\ averaging\ time$$

Equation 4.

$$Potential\ ITSL = \frac{POD}{UF_h \times UF_A \times UF_L} \times \frac{hours\ exposed}{AT}$$

Where:

- POD=point of departure=NOAEL =500 ppm or 1228.9632 mg/m³
- Hours exposed =6 hours per day
- UF_h= uncertainty factor for average human to sensitive human extrapolation=10
- UF_A= uncertainty factor for animal to human extrapolation=10
- UF_L= uncertainty factor for LOAEL to NOAEL extrapolation=1

$$Potential\ ITSL = \frac{1228.9632 \frac{mg}{m^3}}{10 \times 10} \times \frac{6\ hours}{24\ hours\ per\ day} \times 1000 \frac{\mu g}{mg} = 3072.408 \frac{\mu g}{m^3}$$

$$Potential\ ITSL \approx 3000 \frac{\mu g}{m^3},\ 24\ hr\ averaging\ time$$

There are a number of limitations with using this study as the key study for ITSL derivation. While local effects were observed, these were not considered adverse and a LOAEL and NOAEL for local effects was not clearly described in the ECHA summary. As a result, using 500 ppm as the NOAEL is a judgement determination made here. The original research study, which may have more details about the experimental design and the results, could not be obtained. Because of this lack of information, this study will not be used for the final ITSL derivation.

The unadjusted TLV is expected to be health protective for developmental effects, because the TLV documentation states, “Based on comparisons of n-propanol studies in rats, Nelson et al. concluded that n-propanol was neither a selective developmental toxin nor would exposure to this material place human females at risk for alcohol-induced birth defects who are occupationally exposed at concentrations no greater than 200 ppm 8-hour TWA” (Nelson et al., 1990; ACGIH, 2007). While the derivation for the TLV itself is not known, the TLV support documentation further states that the TLV is “based on animal models of sensory irritation and on the structure activity relationship to 2-propanol” (ACGIH, 2007). This has been previously noted as a major limitation of the TLV, since it is based on acute toxicity data and a chemical structure comparison instead of a chemical-specific repeated study (MDNR, 1992). Especially considering the differences in relative potencies and critical effects, use of chemical structure comparisons without clear, detailed documentation of the rationale is a limitation of the TLV. However, protection against the acute effects seems most appropriate given the potential for portal of entry effects, irritancy, and developmental effects. A potential ITSL can be calculated from the TLV as shown in Equation 4 pursuant to AQD Rule 336.1232 (1)(c).

Equation 4.

$$ITSL = \frac{OEL}{100}$$

$$potential\ ITSL = \frac{246 \frac{mg}{m^3}}{100} \times \frac{10^3 \mu g}{mg} = 2,460 \frac{\mu g}{m^3} \approx 2500 \frac{\mu g}{m^3}, 8\ hour\ averaging\ time$$

Comparing the possible potential ITSLs, it may not be appropriate to derive an ITSL from an oral study since irritancy/point of contact effects seem to be the critical effects. Furthermore, based on the weight of evidence from potential ITSLs derived to be protective of local effects, these ITSLs would also be health-protective of potential systemic effects like developmental toxicity. Lastly, while there are limitations to all of the ITSLs based on local effects, an acute ITSL is more appropriate to protect against acute effects like irritation, and the ITSL based on the TLV is more appropriate to use as it is based on identified LOAEL and NOAEL information and has been evaluated multiple times by ACGIH.

Therefore, the ITSL for propyl alcohol is 2500 µg/m³, 8 hour averaging time.

References

Act 451 of 1994, Natural Resources and Environmental Protection Act and Air Pollution Control Rules, Michigan Department of Environmental Quality.

ACGIH. 2001. Documentation of the Threshold Limit Values and Biological Exposure Indices-2-propanol. American Conference of Governmental Industrial Hygienists, Cincinnati, OH.

ACGIH. 2007. Documentation of the Threshold Limit Values and Biological Exposure Indices-n-propanol. American Conference of Governmental Industrial Hygienists, Cincinnati, OH.

ACGIH. 2008. Documentation of the Threshold Limit Values and Biological Exposure Indices-Ethanol. American Conference of Governmental Industrial Hygienists, Cincinnati, OH.

ECB. 2008. European Union Risk Assessment: Propan-1-ol, CAS No. 71-23-8. European Chemicals Bureau, Germany.

ECHA. 2018. REACH Registration Dossier: Propan-1-ol, EC/List. No: 200-746-9, CAS no: 71-23-8. European Chemicals Agency. <https://echa.europa.eu/registration-dossier/-/registered-dossier/14586/7/6/3/?documentUID=e1aaf42e-2f24-46f4-93e0-22a9ba1c983b> Accessed May 2, 2019.

EPA/OTS. 1992. Initial Submission: Experimental Investigation of Carcinogenic Effects of Solvents Exemplified by 1-propanol, 2-methyl-1-propanol & 3-methyl-1-butanol with Cover letter 012092. U.S. EPA Fiche #: OTS0533858.

EPA. 1991. Guidelines for Developmental Toxicity Risk Assessment. Risk Assessment Forum, U.S. Environmental Protection Agency, Washington D.C. EPA/600/FR-91/001.

EPA. 1997. Health Effects Assessment Summary Tables (HEAST). U.S. Environmental Protection Agency, Washington, D.C.

EPA. 2002. A Review of the Reference Dose and Reference Concentration Processes. Risk Assessment Forum, U.S. Environmental Protection Agency, Washington D.C. EPA/630/P-02/002F.

EPA. 2012. Advances in Inhalation Gas Dosimetry for Derivation of a Reference Concentration (RfC) and Use in Risk Assessment. U.S. Environmental Protection Agency, Washington, DC, EPA/600/R-12/044.

EPA. 2007. Provisional Peer Reviewed Toxicity Values for n-propyl alcohol (CASRN 71-23-8). United States Environmental Protection Agency, Office of Superfund Remediation and Technology Innovation, Waste and Cleanup Risk Assessment. Assessed on March 21, 2019 at https://hhprrtv.ornl.gov/issue_papers/PropylAlcoholn.pdf.

Gibel, W., Lohs, K. H., Wildner, G., Schramm, T. 1974. Experimental studies on the carcinogenic effect of higher alcohols using 3-methyl-1-butanol 1-propanol and 2-methylpropanol as examples. Zeitschrift fuer Experimentelle Chirurgie, 7(4), 235-239.

Gibel, W., Lohs, K. H., Wildner, G. P. 1975. Experimental study on cancerogenic activity of Propanol-1, 2-Methylpropanol-1 and 3-Methylbutanol-1. Archiv fur Geschwulstforschung, 45(1), 19-24.

Hazardous Substances Data Bank (HSDB) [Internet]. 2001. Bethesda (MD): National Library of Medicine (US); [Last Revision Date March 2008; cited on March 21, 2019]. N-Propanol; Hazardous Substances Databank Number: 115. Available from: <http://toxnet.nlm.nih.gov/cgi-bin/sis/search2/r?dbs+hsdb:@term+@DOCNO+115>

MDNR. 1992. *Memo from Gary Butterfield to File for n-Propanol (CAS # 71-23-8)*. July 22, 1992. Michigan Department of Natural Resources.

MDEQ. 2016. *Memo from Robert Sills to File for Isopropyl Alcohol (CAS # 67-63-0)*. December 1, 2016. Michigan Department of Environmental Quality, Air Quality Division

MDEQ. 2017. *Memo from Robert Sills to File for Ethyl alcohol (CAS # 64-17-5)*. February 9, 2017. Michigan Department of Environmental Quality, Air Quality Division

EGLE Chemical Criteria Database. N-propyl alcohol (CAS Number 71-23-8). Surface Water Quality Division Reference Dose, calculated on May 14, 1992. Air Quality Division ITSL, calculated on July 22, 1992. Accessed on March 21, 2019.

National Center for Biotechnology Information. PubChem Compound Database; CID=174, <https://pubchem.ncbi.nlm.nih.gov/compound/174> (accessed Sept. 13, 2017).

Nelson, B. K.; Brightwell, W. S.; MacKenzie-Taylor, D. R.; Khan, A.; Burg, J. R.; Weigel, W. W.; Goad, P. T. 1988. Teratogenicity of n-propanol and isopropanol administered at high inhalation concentrations to rats. *Food and chemical toxicology*, 26(3), 247-254.

Nelson, B. K.; Brightwell, W. S.; Krieg Jr, E. F. 1990. Developmental toxicology of industrial alcohols: a summary of 13 alcohols administered by inhalation to rats. *Toxicology and industrial health*, 6(3-4), 373-387.

NIOSH. Pocket Guide to Chemical Hazards: N-Propyl alcohol. Accessed March 21, 2019. The page was last reviewed on November 29, 2018. <https://www.cdc.gov/niosh/npg/npgd0533.html>

OSHA. Occupational Chemical Database: N-Propyl alcohol (N-Propanol). Accessed March 21, 2019. The page was last updated on December 19, 2018. <https://www.osha.gov/chemicaldata/chemResult.html?recNo=489>

TCEQ. 2015. Tox ESL-Summary Report for 1-Propanol. Accessed from the Texas Air Monitoring Information web interface on March 21, 2019. <https://www17.tceq.texas.gov/tamis/index.cfm?fuseaction=report.main>

Michigan Department of Natural Resources

Interoffice Communication

July 22, 1992

To : n-Propanol File (CAS # 71-23-8)

From : Gary Butterfield

Subject : ITSL for n-Propanol

The ACGIH has a TLV of 200 ppm (or 500 mg/m³) for n-propanol. However, it can be recommended that the ITSL not be derived from the TLV, as the TLV is based on acute toxicity data and a chemical structure comparison to other alcohols and their TLV, rather than an identified exposure level with no observed effects from a soundly conducted long term study. A question of confidence in an ITSL based on a TLV derived from such data could be raised.

Few toxicity studies evaluating n-propanol are available. One of the best of the available studies is Nelson et al (1988), who exposed pregnant rats to propanol vapors at concentrations of 0, 3500, 7000 or 10000 ppm during their gestation. Fetal skeletal malformations were increased, and fetal body weights were decreased in fetuses from dams exposed to 7000 and 10000 ppm. From this study a NOAEL of 3500 ppm (or 8750 mg/m³) can be identified. The route of exposure in this study, inhalation, is appropriate for use in development of an ITSL. Although exposure during gestation may be considered to be of a relatively short duration, when compared to lifetime studies, the NOAEL from this study will be used to derive the ITSL. Using the NOAEL from this study to derive the ITSL is appropriate because the NOAEL comes from a well conducted study of a sensitive life stage. The ITSL can be calculated as follows.

$$\text{ITSL} = (8750 \text{ mg/m}^3) / (35 \times 100) \times (7/24) = 730 \text{ ug/m}^3$$

with annual averaging

References :

ACGIH. 1986. Documentation of the TLV and BEI.

Nelson et al. 1988. Teratogenicity of n-propanol and isopropanol at high inhalation concentrations to rats. *Fd Chem Toxicol* 29:247-254.

Nelson et al. 1989. Behavioral teratology investigation of n-propanol administered by inhalation to rats. *Neurotoxicol Teratol* 11:153-159.

Nelson et al. 1990. Developmental toxicology of industrial alcohols: a summary of 13 alcohols administered by inhalation to rats. *Toxicol Ind Health* 6:373-387.